

## IN SILICO DESIGN OF POTENTIAL 1,5-BENZOTHAZEPINE DERIVATIVES AS AN ANTI- CONVULSANT AGENT BY MOLECULAR DOCKING STUDIES

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### ABSTRACT

Epilepsy is characterized by the presence of recurrent seizures. A seizure can be defined as “an episodic disturbance of movement, feeling, or consciousness caused by sudden synchronous, inappropriate, and excessive electrical discharges in the cerebral cortex”. One in every three patients with epilepsy is probable to be severely disabled. It is continuing this scenario as an attempt to develop potent and nontoxic anti-convulsant agents. Recently discovery of benzothiazepine derivatives as an anticonvulsant agent is significant area for research in medicinal chemistry as it is free from all side effects which is shown by a developed as an anticonvulsant agent. In this paper, we have presented results of 2D, and 3D docking poses studies of a series of 300 (Three series) molecules containing 1,5-benzothiazepine pharmacophore as anti-convulsant agents. Docking analysis was utilized to predict the mechanism of action of the designed derivatives for anticonvulsant potential. All the molecules exhibited binding score in the range of -82.61 to -118.25 kcal/mol. Most active molecules from Series 1, 2 and 3 exhibited hydrogen bond interactions with LEU282B, LEU282B and LEU282B. Also for the selected standard sodium phenytoin showed the hydrogen bond interaction with LYS637A. It was noted that the docking score of 1a to 10a, 101b to 110b and 201c to 210c was almost same as that of selected standard sodium phenytoin. Protein showed hydrogen bonding with all synthesized compound showed potential against the epilepsy with GABA nergic mechanism.

**Keywords:** Anti-convulsant; 1,5-benzothiazepine; V-Life MDS 4.3.

### INTRODUCTION

Epilepsy is characterized by the presence of recurrent seizures. A seizure can be defined as “an episodic disturbance of movement, feeling, or consciousness caused by sudden synchronous, inappropriate, and excessive electrical discharges in the cerebral cortex” [1]. Epileptic convulsions are expected to have negative consequences on the patient’s psychological and social life such as relationships, education and employment. Uncontrolled seizures are associated with physical and psychosocial morbidity, dependent behaviour, poor quality of life and an increased risk of sudden unexpected death. Therefore, it is often recommended to begin treatment of epilepsy with antiepileptic drugs (AEDs) as soon as the patient has reported more than one documented or witnessed seizure bearing in mind that the goal of treatment should be to maintain as normal a life style through complete seizure control with no or minimal side effects [2].

Anti-convulsant drugs are widely used in the treatment of various central nervous system diseases like bipolar disorder, antipsychotic, impulsive aggression, borderline personality disorder etc. Benzothiazepine is the most vital class of series origin of benzodiazepine

pharmacophore. They are differ only in place of sulphur and nitrogen element in the heterocyclic ring system. Particularly benzothiazepines used as a cardiovascular-related diseases viz coronary vasodilation, hypertension etc. Recently it has been used as a anti-convulsant, antipsychotic activity, anti-HIV activity, antimicrobial activity etc. There are various benzothiazepines which have been synthesized and tested for their biological activities (3-6). 1,5 benzothiazepine is a calcium channel blocker also known as Diltiazem [3]. Diltiazem is a non-dihydropyridine (DHP) member of the group of drugs known as benzothiazepine.

Medicinal chemists today are facing many complicated challenges. The most demanding and perhaps the most rewarding one is the rational design of new therapeutic agents for treating human diseases. The definition currently accepted of what molecular modelling can be stated as “molecular modelling is anything that requires the use of a computer to paint, describe or evaluate any aspect of the properties of the structure of a molecule”. Methods used in the molecular modelling are regarding automatic structure generation, analysis of three-dimensional (3D) databases and construction of protein models by techniques based on sequence homology, diversity analysis, docking of ligand. Molecular modelling has widened the horizons of pharmaceutical research by providing tools for finding new leads.

Thus, today, molecular modelling is regarded as a field concerned with the use of all sort of different strategies to model and to deduce information of a system at the



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atomic level. On the other hand, this discipline includes all methodologies used in computational chemistry, like computation of the energy of a molecular system, energy minimisation Monte Carlo methods or molecular dynamics. In other words, it is possible to conclude that computational chemistry is the nucleus of molecular modelling. Identification of bio-molecular moieties involved in the interaction with a specific receptor permits to understand the molecular mechanism responsible of its particular biological activity. In turn, this knowledge is aimed at designing new active molecules that can be successfully used as drugs. Because simulation accuracy is limited to the precision of the constructed models, when it is possible, computational simulations have to be compared with experimental results to confirm model accuracy and to modify them if necessary, in order to obtain better representations of the system [4-6].

The developments of new anti-convulsant therapeutic agents are one of the fundamental goals in medicinal chemistry. In recent years there has been concerned search for the discovery and development of potent and selective against anti-convulsant agents. Heterocyclic compounds comprise the dominant family of organic compounds. These are enormously essential with a wide range of synthetic, pharmaceutical and industrial applications and are famous for their biological activities. There is an extensive spectrum of biological activities shown by many compounds containing five-membered heterocyclic rings in their structure.

Therefore we attempt to identify the potential molecule for the synthesis of 1,5-benzothiazepine as an anticonvulsant using V-Life MDS 4.3 software for the execution of synthesis of selected moiety. From the present work, we find out the physicochemical and interactive parameters responsible for the anticonvulsant action of new 1,5-benzothiazepine as anti-convulsant agents from the docking studies,

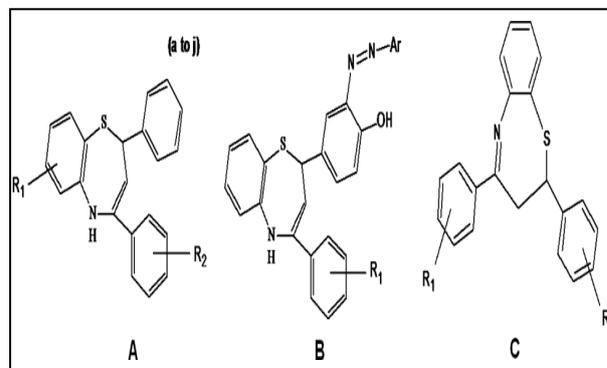
## MATERIAL AND METHODOLOGY

**Equipments:** All computational studies were performed using V-Life Molecular Design Software Version 4.3. Docking study were generated using a training set of 300 molecules.

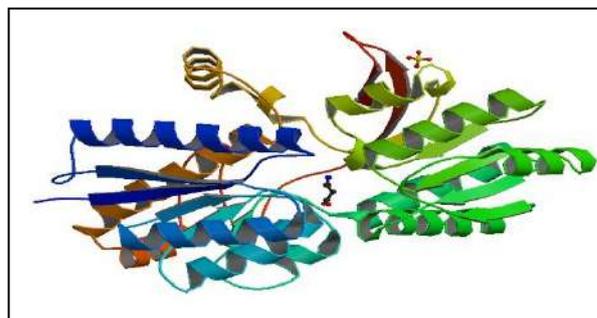
**Molecular docking studies:** V Life MDS version 4.3 software was employed to assess the structure of the enzyme-inhibitor complex. In our study, three series of 1,5 benzothiazepine were selected for docking studies (**Figure 1a, 1b and 1c**) Three hundred structure of 1,5-benzothiazepine expected derivatives were tested and also shown in **Table 1, 2 and 3**. VLifeMDS version 4.3 expected binding free energies of enzyme-inhibitor complexes and the binding energies of both the bound and unbound states using semi-empirical free energy force field. The 3D structures of following PDBs were acquired from RCSB Protein Data Bank. The PDB id is, 3IP9 (Structure of Atu2422-GABA receptor in complex with GABA shown in **Figure 2**) [7]. The 3D structures of selected 1,5-benzothiazepine derivatives were drawn in ACD-Chemsketch and converted into 3D mol. format. The automated docking model was generated using Vlife MDS Tool. The co-crystallized ligand was used to generate the grid box for catalytic inhibition mode. The selected grid box size was 60×60×60.

PyMOL 1.7.4 and LigPlot+ were adopted to acquire the number of H-bonds and van der Waals interacting residues. To determine the binding affinities between the ligand and receptor, the amino acids with the binding pockets was predicted at Q-site finder server [8].

## RESULTS



**Fig 1. Selected 1,5-benzothiazepine series for the docking study**

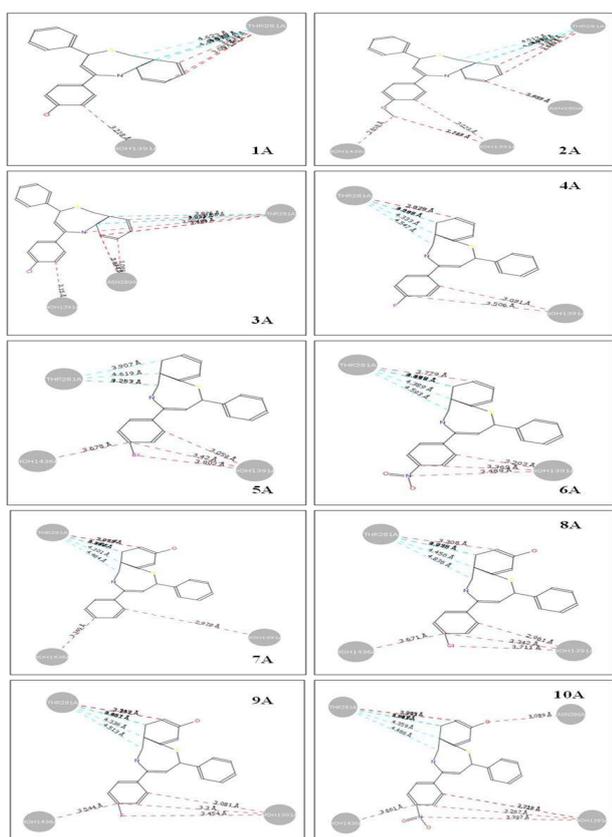


**Fig 2. Target structure PDB code 3IP9 (Structure of Atu2422-GABA receptor in complex with GABA)**

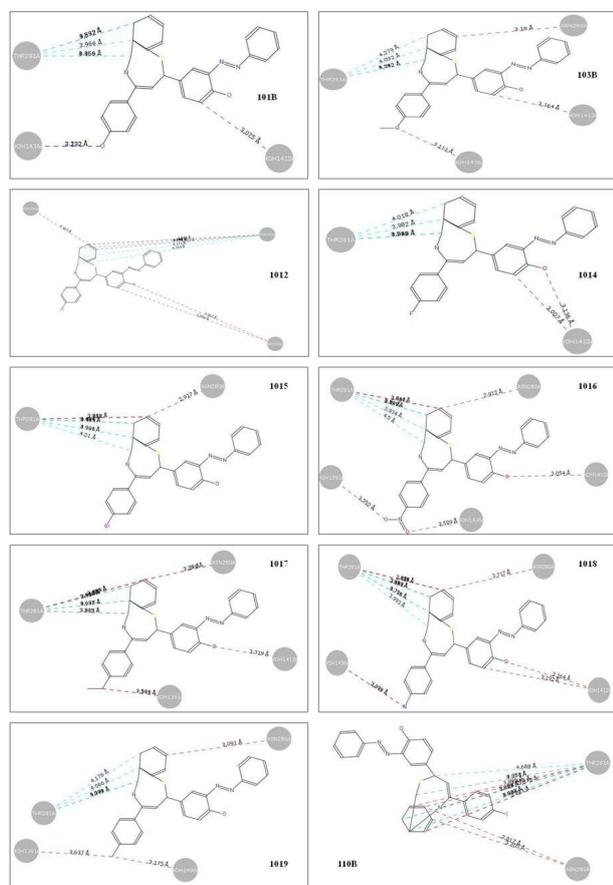
**Docking Study:** Molecular docking approach was utilized to guess the enzyme inhibitor interaction geometries for the selected compounds. The docking scores for 1,5-benzothiazepine derivatives (selected 300 substituents on 1,5-benzothiazepine moieties) with interacting 3IP9 (Structure of Atu2422-GABA receptor in complex with GABA) residues including hydrogen bond, van der Waals and hydrophobic interacting residues. Sodium Phenytoin moiety was chosen as a standard drug for the docking study.

## DISCUSSION

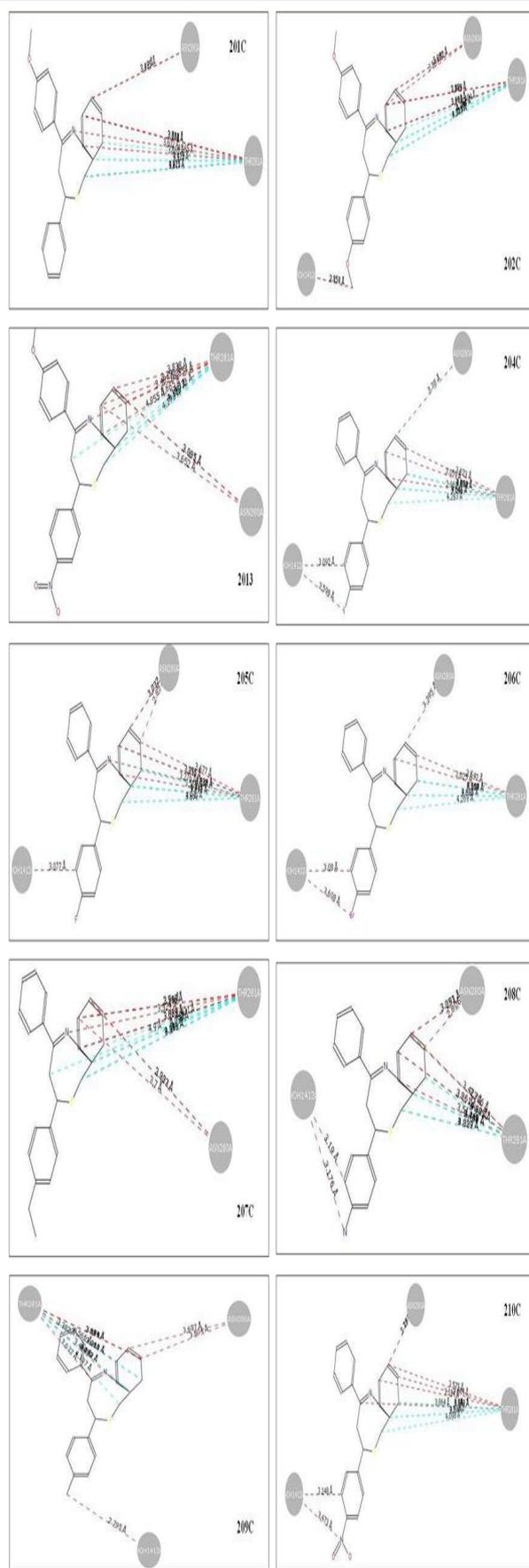
Epilepsy is associated with physical and psychosocial morbidity, dependent behavior, poor quality of life and an increased risk of sudden unexpected death, therefore it is an urgent social need to discover a new potential derivative for the treatment of epilepsy. Henceforth we have planned to synthesize a new derivative from the class of 1,5 benzothiazepine. Generally we have drawn the three series for the synthesis of 1,5 benzothiazepine moiety to get higher potential moiety. Three hundred substituents of the selected series were fixed and were screened to get better activity against the epilepsy. As per the discussion in the introduction part molecular modeling study is the best option to get the idea about the potency of



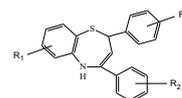
**Fig 3. 2D images of 1a-10a interacting with 3IP9 protein**



**Fig 4. 2D images of 101b-110b bonding with 3IP9 protein**

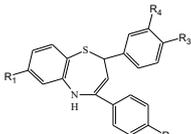


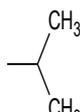
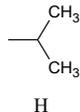
**Fig 5. 2D images of 201C-210C bonding with 3IP9 protein**

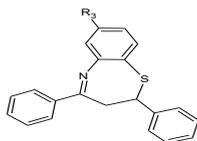
**Table 1. Proposed chemical structure (1-100) for docking study**

Compound Code	R1	R2	R3	Compound Code	R1	R2	R3
1	H	OH	H	51	I	Br	CH <sub>3</sub>
2	H	OCH <sub>3</sub>	H	52	I	Cl	CH <sub>3</sub>
3	H	Cl	H	53	I	NO <sub>2</sub>	CH <sub>3</sub>
4	H	F	H	54	I	NH <sub>2</sub>	CH <sub>3</sub>
5	H	Br	H	55	I	NH <sub>2</sub>	I
6	OH	Br	H	56	I	NH <sub>2</sub>	F
7	OH	Br	OH	57	I	NH <sub>2</sub>	Br
8	H	NO <sub>2</sub>	H	58	I	NH <sub>2</sub>	NO <sub>2</sub>
9	Cl	H	H	59	I	NH <sub>2</sub>	NH <sub>2</sub>
10	F	H	H	60	F	NH <sub>2</sub>	F
11	Br	H	H	61	F	NO <sub>2</sub>	F
12	OH	H	H	62	F	I	F
13	OH	I	H	63	F	Cl	F
14	OH	Cl	H	64	F	Br	F
15	OH	F	H	65	F	NH-CH <sub>3</sub>	F
16	H	NO <sub>2</sub>	H	66	I	NH-CH <sub>3</sub>	F
17	Br	NO <sub>2</sub>	H	67	Br	NH-CH <sub>3</sub>	F
18	F	NO <sub>2</sub>	H	68	H	NH-CH <sub>3</sub>	F
19	NO <sub>2</sub>	NO <sub>2</sub>	H	69	H	NH-CH <sub>3</sub>	H
20	OH	NO <sub>2</sub>	H	70	H	NH-CH <sub>3</sub>	OH
21	OH	NO <sub>2</sub>	Cl	71	H	NH-CH <sub>3</sub>	CH <sub>3</sub>
22	OH	NO <sub>2</sub>	F	72	H	NH-CH <sub>3</sub>	I
23	OH	NO <sub>2</sub>	Br	73	H	NH-CH <sub>3</sub>	Cl
24	OH	NO <sub>2</sub>	NO <sub>2</sub>	74	H	NH-CH <sub>3</sub>	Br
25	OH	NO <sub>2</sub>	I	75	NO <sub>2</sub>	NH-CH <sub>3</sub>	Br
26	OH	H	H	76	NO <sub>2</sub>	NH-CH <sub>3</sub>	F
27	H	H	H	77	NO <sub>2</sub>	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>
28	OH	CH <sub>3</sub>	CH <sub>3</sub>	78	NH <sub>2</sub>	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>
29	CH <sub>3</sub>	H	H	79	F	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>
30	CH <sub>3</sub>	H	Br	80	I	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>
31	CH <sub>3</sub>	H	F	81	I	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	CH <sub>3</sub>
32	CH <sub>3</sub>	H	Cl	82	Cl	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	CH <sub>3</sub>
33	CH <sub>3</sub>	H	CH <sub>3</sub>	83	NO <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	CH <sub>3</sub>
34	CH <sub>3</sub>	Cl	CH <sub>3</sub>	84	NH <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	CH <sub>3</sub>
35	CH <sub>3</sub>	Br	CH <sub>3</sub>	85	NH <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	F
36	CH <sub>3</sub>	F	CH <sub>3</sub>	86	NH <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	Br
37	CH <sub>3</sub>	NO <sub>2</sub>	CH <sub>3</sub>	87	Cl	C=O	H
38	CH <sub>3</sub>	NO <sub>2</sub>	H	88	H	C=O	H
39	CH <sub>3</sub>	NO <sub>2</sub>	Cl	89	Br	C=O	H
40	CH <sub>3</sub>	NO <sub>2</sub>	F	90	OH	C=O	H
41	CH <sub>3</sub>	NO <sub>2</sub>	Br	91	I	C=O	H
42	CH <sub>3</sub>	CH <sub>3</sub>	H	92	NO <sub>2</sub>	C=O	H
43	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>	93	F	C=O	H
44	Cl	CH <sub>3</sub>	CH <sub>3</sub>	94	NH <sub>2</sub>	C=O	H
45	Br	CH <sub>3</sub>	CH <sub>3</sub>	95	NH <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	I
46	F	CH <sub>3</sub>	CH <sub>3</sub>	96	NH <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	NO <sub>2</sub>
47	NO <sub>2</sub>	CH <sub>3</sub>	CH <sub>3</sub>	97	NO <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	I
48	I	CH <sub>3</sub>	CH <sub>3</sub>	98	NO <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	Br
49	I	I	CH <sub>3</sub>	99	NO <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	NH <sub>2</sub>
50	I	F	CH <sub>3</sub>	100	NO <sub>2</sub>	$\begin{array}{c} \text{CH}_3 \\   \\ \text{-HC-CH}_3 \end{array}$	NO <sub>2</sub>

Table 2. Proposed chemical structure (101-200) for docking study



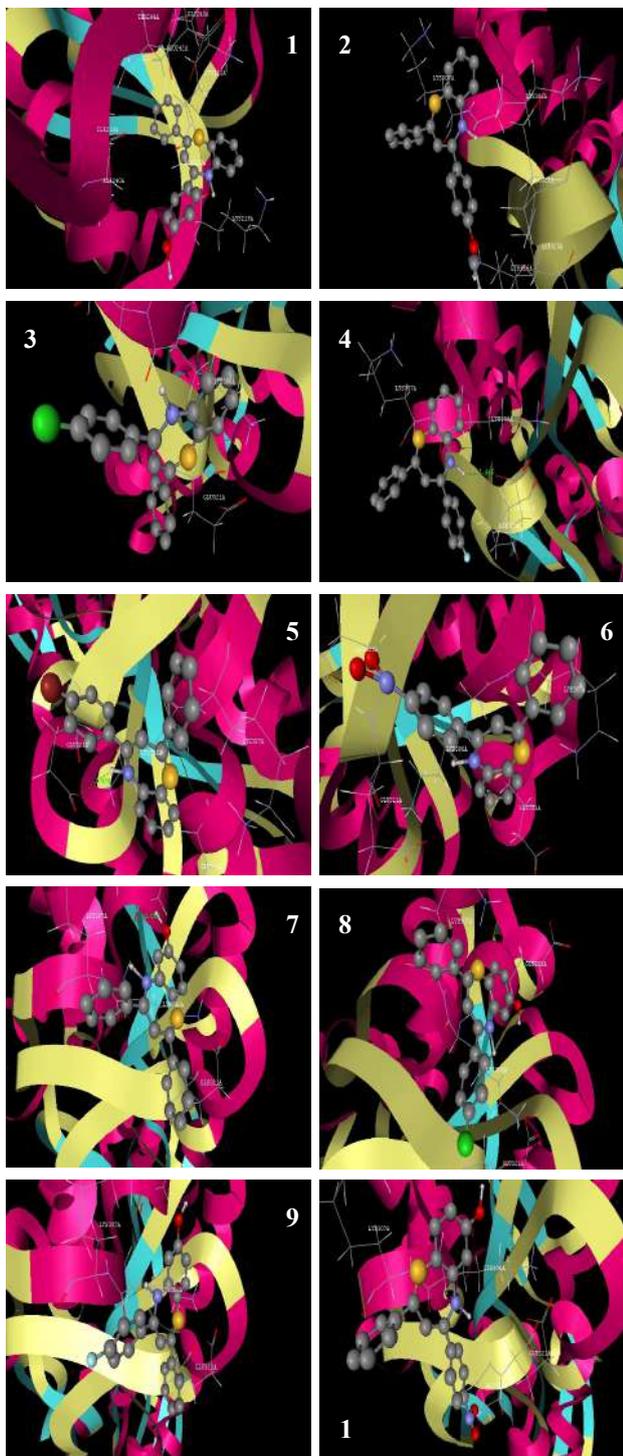
Compound Code	R1	R2	R3	R4	Compound Code	R1	R2	R3	R4
101	H	OH	OH		151	H	F	CH <sub>3</sub>	NH <sub>2</sub>
102	H	OCH <sub>3</sub>	OH		152	H	Cl	CH <sub>3</sub>	NH <sub>2</sub>
103	H	Cl	OH		153	H	Br	CH <sub>3</sub>	NH <sub>2</sub>
104	H	I	OH		154	H	NO <sub>2</sub>	CH <sub>3</sub>	NH <sub>2</sub>
105	H	F	OH		155	H	NH <sub>2</sub>	CH <sub>3</sub>	NH <sub>2</sub>
106	H	-CH <sub>2</sub> -CH <sub>3</sub>	OH		156	H	CH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
107	H	Br	OH		157	H	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
108	H	NO <sub>2</sub>	OH		158	OH	NH-CH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
109	H	NH <sub>2</sub>	OH		159	H	NH-CH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
110	H	CH <sub>3</sub>	OH		160	OH	NH-I	CH <sub>3</sub>	NH <sub>2</sub>
111	OH	Cl	OH		161	OH	NH-F	CH <sub>3</sub>	NH <sub>2</sub>
112	H	CH <sub>3</sub>	Br		162	OH	NH-Br	CH <sub>3</sub>	NH <sub>2</sub>
113	H	NH <sub>2</sub>	Br	NH <sub>2</sub>	163	OH	I	NH <sub>2</sub>	NH <sub>2</sub>
114	H	NO <sub>2</sub>	Br	NH <sub>2</sub>	164	OH	F	NH <sub>2</sub>	NH <sub>2</sub>
115	H	Br	Br	NH <sub>2</sub>	165	OH	Cl	NH <sub>2</sub>	NH <sub>2</sub>
116	H	Cl	Br	NH <sub>2</sub>	166	OH	Br	NH <sub>2</sub>	NH <sub>2</sub>
117	H	F	Br	NH <sub>2</sub>	167	OH	CH <sub>3</sub>	NH <sub>2</sub>	NH <sub>2</sub>
118	H	I	Br	NH <sub>2</sub>	168	OH	NO <sub>2</sub>	NH <sub>2</sub>	NH <sub>2</sub>
119	H	I	NO <sub>2</sub>	NH <sub>2</sub>	169	OH	NH <sub>2</sub>	NH <sub>2</sub>	NO <sub>2</sub>
120	H	F	NO <sub>2</sub>	NH <sub>2</sub>	170	OH	NH-CH <sub>3</sub>	NH <sub>2</sub>	NO <sub>2</sub>
121	H	Cl	NO <sub>2</sub>	NH <sub>2</sub>	171	OH	NH-CH <sub>3</sub>	NO <sub>2</sub>	NO <sub>2</sub>
122	H	Br	NO <sub>2</sub>	NH <sub>2</sub>	172	OH	CH <sub>2</sub> -CH <sub>3</sub>	NO <sub>2</sub>	NO <sub>2</sub>
123	H	CH <sub>3</sub>	Cl	NH <sub>2</sub>	173	OH	CH <sub>3</sub>	NO <sub>2</sub>	NO <sub>2</sub>
124	H	NH <sub>2</sub>	Cl	NH <sub>2</sub>	174	OH	NH <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>
125	H	NO <sub>2</sub>	Cl	NH <sub>2</sub>	175	OH	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>
126	H	Br	Cl	NH <sub>2</sub>	176	OH	CH <sub>3</sub>		H
127	H	Cl	Cl	NH <sub>2</sub>	177	OH			H
128	H	F	Cl	NH <sub>2</sub>	178	OH	-CH <sub>2</sub> -CH <sub>3</sub>		H
129	H	I	Cl	NH <sub>2</sub>	179	OH	-CH <sub>2</sub> -NH <sub>2</sub>		H
130	H	I	F	NH <sub>2</sub>	180	OH	NH-CH <sub>3</sub>		NH <sub>2</sub>
131	H	F	F	NH <sub>2</sub>	181	OH	NH <sub>2</sub>		H
132	H	Cl	F	NH <sub>2</sub>	182	OH	Br		H
133	H	Br	F	NH <sub>2</sub>	183	OH	Cl		H
134	H	NO <sub>2</sub>	F	NH <sub>2</sub>	184	OH	F		H
135	H	NH <sub>2</sub>	F	NH <sub>2</sub>	185	OH	I		H
136	H	CH <sub>3</sub>	F	NH <sub>2</sub>	186	OH	I		NH <sub>2</sub>
137	H	-CH <sub>2</sub> -CH <sub>3</sub>	I	NH <sub>2</sub>	187	NH <sub>2</sub>	I	H	H
138	H	CH <sub>3</sub>	I	NH <sub>2</sub>	188	NH <sub>2</sub>	F	H	H
139	H	NO <sub>2</sub>	I	NH <sub>2</sub>	189	NH <sub>2</sub>	NH <sub>2</sub>		H
140	H	NH <sub>2</sub>	I	NH <sub>2</sub>	190	H	NH-CH <sub>3</sub>		NH <sub>2</sub>
141	H	Br	I	NH <sub>2</sub>	191	CH <sub>3</sub>	NH <sub>2</sub>		NH <sub>2</sub>
142	H	Cl	I	NH <sub>2</sub>	192	CH <sub>3</sub>	NH-CH <sub>3</sub>	H	NH <sub>2</sub>
143	H	F	I	NH <sub>2</sub>	193	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
144	H	CH <sub>2</sub> I	I		194	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>2</sub> Cl	NH <sub>2</sub>
145	H	CH <sub>2</sub> I	OH		195	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>2</sub> Br	NH <sub>2</sub>
146	H	CH <sub>2</sub> F	OH		196	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>2</sub> I	NH <sub>2</sub>
147	H	CH <sub>2</sub> Cl	OH		197	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>2</sub> NH <sub>2</sub>	NH <sub>2</sub>
148	H	CH <sub>2</sub> Br	OH		198	CH <sub>3</sub>	NH-CH <sub>3</sub>	CH <sub>2</sub> NO <sub>2</sub>	NH <sub>2</sub>
149	H	CH <sub>2</sub> NO <sub>2</sub>	OH	NH <sub>2</sub>	199	Cl	NH <sub>2</sub>	H	NH <sub>2</sub>
150	H	CH <sub>2</sub> NH <sub>2</sub>	OH	NH <sub>2</sub>	200	Cl	NH-NH <sub>2</sub>	CH <sub>2</sub>	NH <sub>2</sub>

**Table 3. Proposed chemical structure (201-300) for docking study**

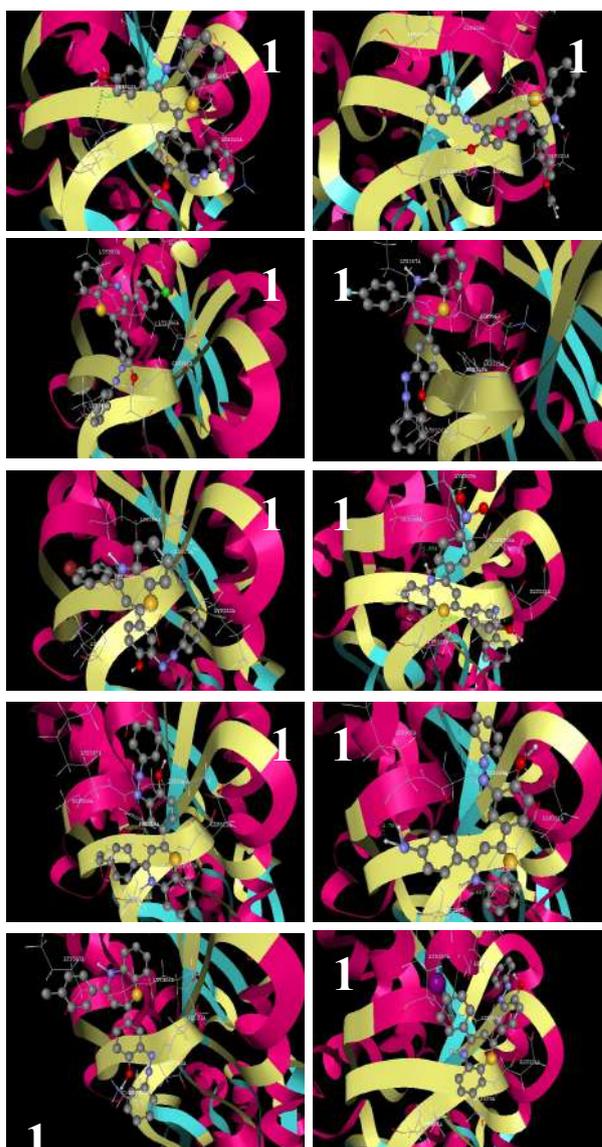
Compound Code	R1	R2	R3	Compound Code	R1	R2	R3
201	OCH <sub>3</sub>	H	H	251	CH <sub>3</sub>	H	NH <sub>2</sub>
202	OCH <sub>3</sub>	OCH <sub>3</sub>	H	252	CH <sub>3</sub>	H	CH <sub>3</sub>
203	OCH <sub>3</sub>	NO <sub>2</sub>	H	253	CH <sub>3</sub>	H	CH <sub>2</sub> -CH <sub>3</sub>
204	H	I	H	254	CH <sub>3</sub>	H	CH <sub>2</sub> -NO <sub>2</sub>
205	H	F	H	255	CH <sub>3</sub>	H	CH <sub>2</sub> -NH <sub>2</sub>
206	H	Br	H	256	CH <sub>3</sub>	H	H
207	H	NO <sub>2</sub>	H	257	CH <sub>3</sub>	H	OH
208	H	NH <sub>2</sub>	H	258	CH <sub>3</sub>	H	NH <sub>2</sub>
209	H	CH <sub>3</sub>	H	259	H	H	CH <sub>3</sub>
210	H	CH <sub>2</sub> -CH <sub>3</sub>	H	260	OH	H	CH <sub>3</sub>
211	I	CH <sub>2</sub> -CH <sub>3</sub>	H	261	CH <sub>3</sub>	NH <sub>2</sub>	H
212	I	CH <sub>3</sub>	H	262	CH <sub>3</sub>	NO <sub>2</sub>	H
213	I	NO <sub>2</sub>	H	263	CH <sub>3</sub>	Br	H
214	I	NH <sub>2</sub>	H	264	CH <sub>3</sub>	Cl	H
215	I	Br	H	265	CH <sub>3</sub>	F	H
216	I	Cl	H	266	CH <sub>3</sub>	I	H
217	I	F	H	267	CH <sub>3</sub>	H	I
218	I	I	H	268	CH <sub>3</sub>	H	Br
219	F	I	H	269	CH <sub>3</sub>	H	F
220	F	F	H	270	CH <sub>3</sub>	H	NO <sub>2</sub>
221	F	Cl	H	271	NH <sub>2</sub>	NO <sub>2</sub>	H
222	F	Br	H	272	NH <sub>2</sub>	NH <sub>2</sub>	H
223	F	NO <sub>2</sub>	H	273	NH <sub>2</sub>	CH <sub>3</sub>	H
224	F	NH <sub>2</sub>	H	274	NH <sub>2</sub>	CH <sub>2</sub> -CH <sub>3</sub>	H
225	F	CH <sub>3</sub>	H	275	NH <sub>2</sub>	CH <sub>2</sub> -NH <sub>2</sub>	H
226	F	CH <sub>2</sub> -CH <sub>3</sub>	H	276	NH <sub>2</sub>	CH <sub>2</sub> -NO <sub>2</sub>	H
227	Cl	CH <sub>2</sub> -CH <sub>3</sub>	H	277	CH <sub>3</sub>	CH <sub>2</sub> -NO <sub>2</sub>	H
228	Cl	CH <sub>2</sub> -NH <sub>2</sub>	H	278	CH <sub>3</sub>	CH <sub>2</sub> -NH <sub>2</sub>	H
229	Cl	CH <sub>3</sub>	H	279	CH <sub>3</sub>	CH <sub>2</sub> -CH <sub>2</sub> -NH <sub>2</sub>	H
230	Cl	NH <sub>2</sub>	H	280	CH <sub>3</sub>	CH <sub>3</sub>	H
231	Cl	NO <sub>2</sub>	H	281	NO <sub>2</sub>	I	H
232	Cl	Br	H	282	NO <sub>2</sub>	F	H
233	Cl	Cl	H	283	NO <sub>2</sub>	Cl	H
234	Cl	F	H	284	NO <sub>2</sub>	Br	H
235	Cl	I	H	285	NO <sub>2</sub>	NH <sub>2</sub>	H
236	Br	I	H	286	NO <sub>2</sub>	NO <sub>2</sub>	H
237	Br	F	H	287	NH <sub>2</sub>	I	CH <sub>3</sub>
238	Br	NH <sub>2</sub>	H	288	NH <sub>2</sub>	F	CH <sub>3</sub>
239	Br	Cl	H	289	NH <sub>2</sub>	Cl	CH <sub>3</sub>
240	Br	Br	H	290	NH <sub>2</sub>	Br	CH <sub>3</sub>
241	Br	NO <sub>2</sub>	H	291	Br	CH <sub>2</sub> -NO <sub>2</sub>	CH <sub>3</sub>
242	Br	OCH <sub>3</sub>	H	292	Br	CH <sub>2</sub> -NH <sub>2</sub>	CH <sub>3</sub>
243	Br	CH <sub>2</sub> -CH <sub>3</sub>	H	293	NO <sub>2</sub>	CH <sub>2</sub> -NO <sub>2</sub>	CH <sub>3</sub>
244	OCH <sub>3</sub>	CH <sub>2</sub> -CH <sub>3</sub>	H	294	NO <sub>2</sub>	CH <sub>2</sub> -NH <sub>2</sub>	CH <sub>3</sub>
245	OCH <sub>3</sub>	NO <sub>2</sub>	H	295	NO <sub>2</sub>	CH <sub>3</sub>	CH <sub>3</sub>
246	OCH <sub>3</sub>	CH <sub>3</sub>	H	296	NO <sub>2</sub>	CH <sub>2</sub> -CH <sub>2</sub> -NH <sub>2</sub>	CH <sub>3</sub>
247	OCH <sub>3</sub>	F	H	297	Br	CH <sub>2</sub> -CH <sub>3</sub>	CH <sub>3</sub>
248	OCH <sub>3</sub>	I	H	298	Br	CH <sub>3</sub>	CH <sub>3</sub>
249	OCH <sub>3</sub>	Cl	H	299	Br	NH <sub>2</sub>	CH <sub>3</sub>
250	OCH <sub>3</sub>	H	H	300	Br	NO <sub>2</sub>	CH <sub>3</sub>

selected derivatives.

A conserved mechanism of GABA binding and antagonism is revealed by Structure-Function Analysis of the periplasmic binding protein Atu2422 [7]. Bacterial periplasmic binding proteins (PBPs) and eukaryotic PBP-like domains (also called as Venus flytrap modules) of G-protein-coupled receptors are involved in extracellular GABA perception (8). Gamma - Aminobutyric acids (GABA), the prime inhibitory neurotransmitter in the cerebral cortex, sustain the inhibitory tenor that counterbalance neuronal excitation



**Fig 6. Docking 3D Images of 1A to 10A synthesized compounds**



**Fig 7. Docking 3D Images of 101B to 110B synthesized compounds**

(9). Therefore we have targeted protein i.e 3IP9 (Structure of Atu2422-GABA receptor in complex with GABA) is particularly found in GABA receptor as per the report of Planamente et al., (2010) [7].

For checking binding interaction first open receptor in MDS followed by compound which saved as ligand dock file. From tool option click on merge molecule so that compound and receptor is merged together. From biopredicta tool edit this complex and select ligand and receptor structure afterward check its interaction. The said protein is correlates with anticonvulsant activity. To discuss the interaction details of ligands and protein, the docked conformations of the compounds were studied on 3IP9. The said proteins belong to the class of GABA receptor group of proteins, which exhibited structural stability to the gene and expression by epigenetic modification of DNA, plays significant role in epilepsy. It was noted that the docking score of 1a to 101a, 101b to 200b and 201c to 300c was almost same as that of selected standard sodium phenytoin. Protein showed hydrogen bonding with some of the compound from selected series with GABAnergic mechanism.

Docking analysis was utilized to predict the mechanism of action of the designed derivatives for anticonvulsant potential. All the molecules exhibited binding score in the range of -82.61 to -118.25 kcal/mol. Most active molecules from Series 1, 2 and 3 exhibited hydrogen bond interactions with LEU282B, LEU282B and LEU282B respectively. Also for the selected standard sodium phenytoin showed the hydrogen bond interaction with LYS637A. The series 1 showed hydrophobic bonding interactions with VAL290B, VAL290B, GLY421B, LEU282B etc. and Vander Waals interactions with LEU282B, LEU282B, GLY283B, PHE 287B, PHE420B (**Figure 3**). The series 2 compound showed hydrophobic bond interactions with GLY421B, PHE420B, VAL 290B, LYS 309B and also Van der Waals interactions with LYS439B, PHE287B, GLY283B, LEU282B, GLY515B (**Figure 4**). The compound from series 3 showed hydrophobic interaction with VAL291B, VAL295B, GLY427B, LEU280B and Van der Waals interactions with LEU284B, LEU285B, GLY280B, PHE207B and PHE421B (**Figure 5**). Whereas standard sodium phenytoin exhibited hydrophobic interaction with ALA549A (4.675), ASP593A (2.987), LYS647A (2.638) and Van der Waals interactions with SER528A (4.658), THR652A (1.675), GLU428A (3.293) and ASN55AA 2.105.

From the series 1, ten derivatives are selected for the synthesis of 1,5 benzothiazepine class namely 1a to 10a. It was observed that the good docking score (**Table 4**). The docking poses with 3D picture are depicted in **Figure 5**. Whereas series 2 have also selected 10 derivative to synthesize the compounds namely 101b to 110b. 3D docking images are depicted in **Figure 6**. The best docking poses of the series 3 of 1, 5 benzothiazepine have been showed in the **Figure 7**. Total 30 compound are having good score which have been selected for the synthesis namely 1a-10a, 101b-110b and 201c to 210c. In the docking study their van. It was discovered that the potential of 1a-100a, 101b-200b, 201c-300c and Sodium Phenytoin against 3IP9 (Structure of Atu2422-GABA receptor in complex with GABA) was linked with the binding energy and the number of bonds formed at the catalytic site. It was noted that the hydrogen bond and hydrophobic interaction of 1a-100a, 101b-200b, 201c-300c and Sodium Phenytoin with proteins further stabilized the enzyme-inhibitor interaction. The 3IP9 had interaction with 1a-100a, 101b-200b, 201c-300c and Sodium Phenytoin through hydrogen bond, the hydrogen bond interaction of 1a-100a, Sodium Phenytoin with 3IP9 proteins further stabilized the enzyme-inhibitor interaction due to amino acid.

## CONCLUSION

In this present investigation, it was found that all projected moiety were statistically significant, therefore from above 2D/3D models it could be concluded that 1,5 benzothiazepine derivatives are used to synthesize as anti-convulsant drugs. It was found that 1a-10a, 101b-110b and 201c-210c having good docking score for synthesizing the derivatives. It was noted that the docking score of 1a to 10a, 101b to 110b and 201c to 230c was almost same as that of selected standard

sodium phenytoin. Protein showed hydrogen bonding with all synthesized compound showed potential against the epilepsy with GABAergic mechanism.

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